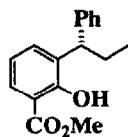


STEREOCHEMISTRY ABSTRACTS

R. Irie, K. Noda, Y. Ito, N. Matsumoto, and T. Katsuki

Tetrahedron: Asymmetry 1991, 2, 481



E.e. = >99 % [by HPLC analysis]

$[\alpha]_D^{25} -180$ (c 1.19, CH₃OH)

Source of chirality: resolution of a precursor

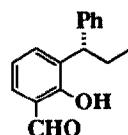
Absolute configuration: *S*

C₁₇H₁₈O₃

Methyl 3-[(*S*)-1-phenylpropyl]salicylate

R. Irie, K. Noda, Y. Ito, N. Matsumoto, and T. Katsuki

Tetrahedron: Asymmetry 1991, 2, 481



E.e. = >99 % [by HPLC analysis of a precursor]

$[\alpha]_D^{21} -257$ (c 1.12, C₂H₅OH)

Source of chirality:methyl 3-[(*S*)-1-Phenylpropyl]salicylate

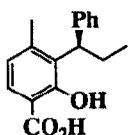
Absolute configuration: *S*

C₁₆H₁₆O₂

3-[(*S*)-1-Phenylpropyl]salicylaldehyde

R. Irie, K. Noda, Y. Ito, N. Matsumoto, and T. Katsuki

Tetrahedron: Asymmetry 1991, 2, 481



E.e. = >99 % [by HPLC analysis of the corresponding methyl ester]

$[\alpha]_D^{24} +33.6$ (c 1.01, C₂H₅OH)

Source of chirality: resolution with (-)brucine

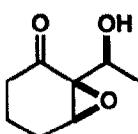
Absolute configuration: *R*

C₁₇H₁₈O₃

4-Methyl-3-[(*R*)-1-phenylpropyl]salicylic acid

M. Bailey, I. Staton, P. R. Ashton, I. E. Markó and W. D. Ollis

Tetrahedron: Asymmetry 1991, 2, 495



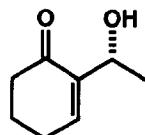
E.e. = 78% [by nmr using Eu(hfc)₃]

$[\alpha]_D^{22} = +37.5$ (c 0.10, CHCl₃)

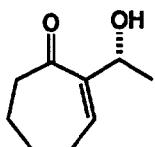
Source of chirality: kinetic resolution [Ti(OPr)₄] / (-) - DET]

C₈H₁₂O₃

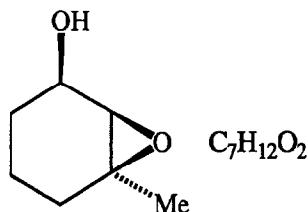
Syn-2-(1-hydroxyethyl)-2,3-epoxy-2-cyclohexan-1-one



E.e. = 88% [by HPLC using a Pirkle 1A]

 $[\alpha]_D^{22} = -5.6$ (c 5.17, CHCl₃)Source of chirality: kinetic resolution [Ti(OPr)₄] / (-) - DETC₈H₁₂O₂ 2-(1-hydroxyethyl)-2-cyclohexen-1-one

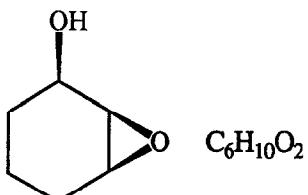
E.e. = 75% [by HPLC using a Pirkle 1A]

 $[\alpha]_D^{22} = -17.6$ (c 0.17, CHCl₃)Source of chirality: kinetic resolution [Ti(OPr)₄] / (-) - DETC₉H₁₄O₂ 2-(1-hydroxyethyl)-2-cyclohepten-1-oneE.e. = 86% (by ¹⁹F nmr of MTPA ester). $[\alpha]_D^{22} = +63.0$ (c = 0.7, CHCl₃).

Source of Chirality : Double Sharpless Epoxidation.

Absolute Stereochemistry : 1R, 2R, 3S

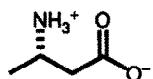
2,3-epoxy-3-methylcyclohexan-1-ol

E.e. = 70% (by ¹H nmr and HPLC of MTPA ester).

Source of Chirality : Double Sharpless Epoxidation.

Absolute Stereochemistry : 1R, 2R, 3S

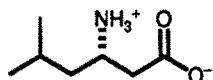
2,3-epoxycyclohexan-1-ol



E.e. = 96% [by HPLC analysis of tetraacetylglucose thiourea derivative]

 $[\alpha]_D^{26} +34.3$ (*c* 1.12, H_2O)Source of chirality: (*R*)-BINAP—Ru(II)-based asymmetric hydrogenationAbsolute configuration: 3*S* $\text{C}_4\text{H}_9\text{NO}_2$

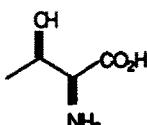
(S)-(+)-3-Aminobutanoic Acid



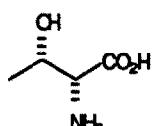
E.e. = 90% [by HPLC analysis of tetraacetylglucose thiourea derivative]

 $[\alpha]_D^{25} +26.7$ (*c* 0.6, H_2O)Source of chirality: (*R*)-BINAP—Ru(II)-based asymmetric hydrogenationAbsolute configuration: 3*S* $\text{C}_7\text{H}_{15}\text{NO}_2$

(S)-(+)-5-Methyl-3-aminohexanoic Acid

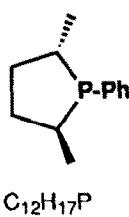


E.e.= 95% (by H.P.L.C. analysis)

Absolute configuration : 2*S*, 3*R*Source of chirality : asymmetric hydrogenation with
(+)-BINAP Ruthenium complexe $\text{C}_4\text{H}_9\text{O}_3$
L Threonine

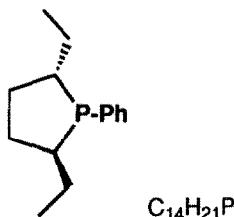
E.e.= 99% (by H.P.L.C. analysis)

Absolute configuration : 2*R*, 3*S*Source of chirality : asymmetric hydrogenation with
(-)CHIRAPHOS Ruthenium complexe $\text{C}_4\text{H}_9\text{O}_3$
D Threonine



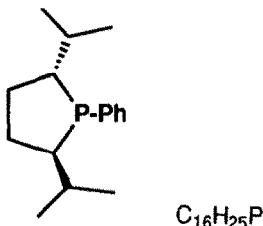
ee = >98% (optical rotation and ^{31}P NMR on chiral Pd complex)
 $[\alpha]_D^{25} = +51.6 \pm 1.0$ (c 1, hexane)
 Source of chirality: Asymmetric hydrogenation
 Absolute configuration: 2S, 5S

2,5-dimethyl-1-phenylphospholane



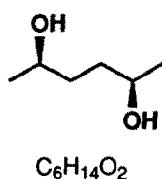
ee = >98% (optical rotation and ^{31}P NMR on chiral Pd complex)
 $[\alpha]_D^{25} = -53.0 \pm 1.0$ (c 1, hexane)
 Source of chirality: Asymmetric hydrogenation
 Absolute configuration: 2S, 5S

2,5-diethyl-1-phenylphospholane



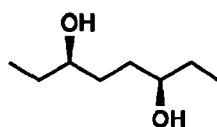
ee = >98% (optical rotation and ^{31}P NMR on chiral Pd complex)
 $[\alpha]_D^{25} = -92.6 \pm 1.0$ (c 1, hexane)
 Source of chirality: Asymmetric hydrogenation
 Absolute configuration: 2R, 5R

2,5-diisopropyl-1-phenylphospholane



ee = >99% (optical rotation and GC on MPTA esters)
 $[\alpha]_D^{25} = -39.6 \pm 0.5$ (c 1, CHCl_3)
 Source of chirality: Asymmetric hydrogenation
 Absolute configuration: 2R, 5R

2,5-hexanediol

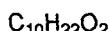
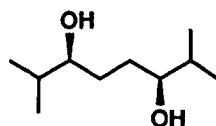


ee = >99% (optical rotation and GC on MPTA esters)

[α]_D²⁵ = -22.8 ± 0.5 (c 1, CHCl₃)

Source of chirality: Asymmetric hydrogenation

Absolute configuration: 3*R*, 6*R*

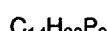
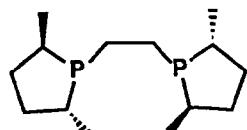
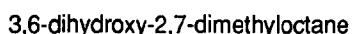


ee = >99% (optical rotation and GC on MPTA esters)

[α]_D²⁵ = -35.2 ± 0.5 (c 1, CHCl₃)

Source of chirality: Asymmetric hydrogenation

Absolute configuration: 3*S*, 6*S*

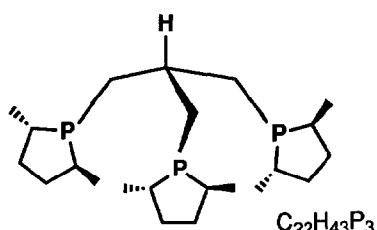


ee = >98% (optical rotation and ³¹P NMR on chiral Pd complex)

[α]_D²⁵ = +263 ± 3.0 (c 1, hexane)

Source of chirality: Asymmetric hydrogenation

Absolute configuration: 2*R*, 5*R* (phospholane)



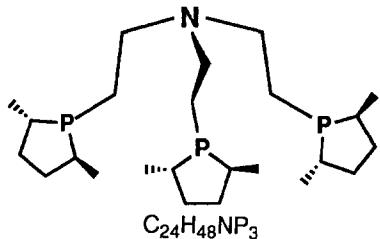
ee = >98% (optical rotation and ³¹P NMR on chiral Pd complex)

[α]_D²⁵ = -329 ± 4.0 (c 1, hexane)

Source of chirality: Asymmetric hydrogenation

Absolute configuration: 2*S*, 5*S* (phospholane)





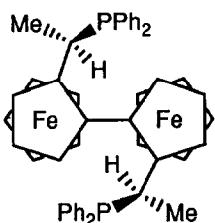
tris((2,5-dimethylphospholano)ethyl)amine

ee = >98% (optical rotation and ^{31}P NMR on chiral Pd complex)

$[\alpha]_D^{25} = -167 \pm 3.0$ (c 1, hexane)

Source of chirality: Asymmetric hydrogenation

Absolute configuration: 2*S*, 5*S* (phospholane)

 $C_{48}H_{44}Fe_2P_2$ (S,S)-2,2''-Bis[(*R*)-1-(diphenylphosphino)ethyl]-1,1''-biferrocene

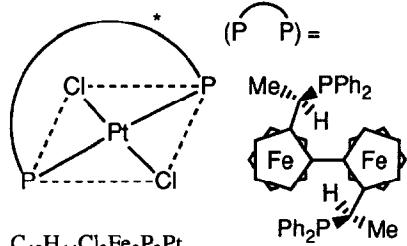
E.e. = 100%

$[\alpha]_D^{25} = -426$ (c 0.51, $CHCl_3$)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R,R*)-(S,S)

mp 99 - 103°C

trans-Dichloro [(S,S)-2,2''-bis[(*R*)-1-(diphenylphosphino)ethyl]-1,1''-biferrocene} platinum(II)

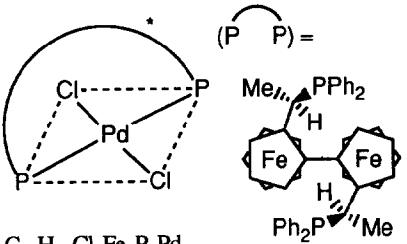
E.e. = 100%

$[\alpha]_D^{20} = -571$ (c 0.57, $CHCl_3$)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R,R*)-(S,S)

mp 240 - 245°C (dec)

trans-Dichloro [(S,S)-2,2''-bis[(*R*)-1-(diphenylphosphino)ethyl]-1,1''-biferrocene} palladium(II)

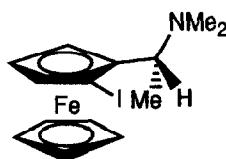
E.e. = 100%

$[\alpha]_D^{20} = -726$ (c 0.55, $CHCl_3$)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R,R*)-(S,S)

mp 230 - 235°C (dec)



Mixture of (*R*)-(S) and (*R*)-(R) [*(R*)-(S) / (*R*)-(R) = 10 / 1]

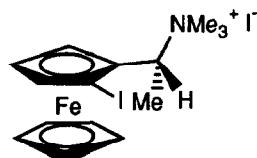
E.e. = 100%

$[\alpha]_D^{25} = -5.4$ (*c* 1.08, CHCl₃)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R*)-(S)

C₁₄H₁₈FeIN (*R*)-*N,N*-Dimethyl-1-[(*S*)-2-iodoferrocenyl]ethylamine



Mixture of (*R*)-(S) and (*R*)-(R) [*(R*)-(S) / (*R*)-(R) = 10 / 1]

E.e. = 100%

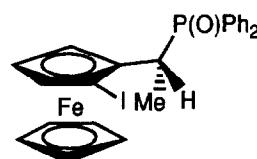
$[\alpha]_D^{25} = -13.0$ (*c* 1.05, CH₃CN)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R*)-(S)

mp 110 - 112°C (dec)

C₁₅H₂₁FeI₂N (*R*)-*N,N,N*-Trimethyl-1-[(*S*)-2-iodoferrocenyl]ethylammonium iodide



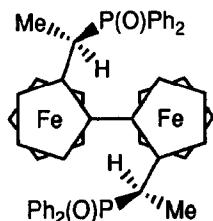
E.e. = 100%

$[\alpha]_D^{25} = +17.9$ (*c* 0.50, CHCl₃)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R*)-(S)

C₂₄H₂₂FeIOP (*S*)-2-[(*R*)-1-(Diphenylphosphinyl)ethyl]-1-iodoferrocene



E.e. = 100% (by HPLC with chiral stationary phase)

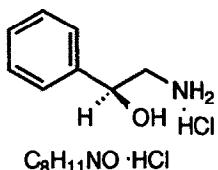
$[\alpha]_D^{25} = -130$ (*c* 1.02, CHCl₃)

Source of chirality: synthesized from (*R*)-*N,N*-Dimethyl-1-(1-ferrocenyl)ethylamine

Absolute configuration: (*R,R*)-(S,S)

mp 245 - 250°C (dec)

C₄₈H₄₄Fe₂O₂P₂ (*S,S*)-2,2''-Bis[*(R*)-1-(diphenylphosphinyl)ethyl]-1,1''-biferrocene

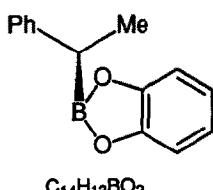


E.e. = 81.0% [by HPLC analysis of free amine]

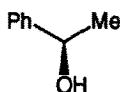
Source of chirality: (2*S*,4*S*)-BCPM-Rh(I)-based asymmetric hydrogenation

Absolute configuration: *S*

1-Amino-2-phenyl-ethanol hydrochloride



E.e. = 93.5% [by converting into

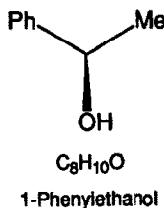


and HPLC analysis of its 3,5-dinitrophenyl carbamate with chiral stationary phase column, Sumipax OA-4100]

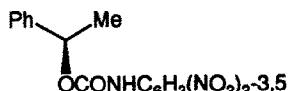
$[\alpha]_D^{20} -55.6$ (*c* 1.7, benzene)

Source of chirality: catalytic asymmetric hydroboration of styrene

Absolute configuration: *R* (oxidized into (*R*)-1-phenylethanol)



E.e. = 96.2% [by converting into

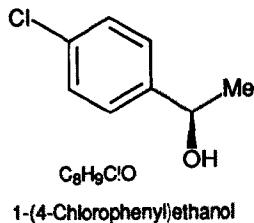


and HPLC with chiral stationary phase column, Sumipax OA-4100]

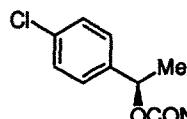
$[\alpha]_D^{23} +48.6$ (*c* 1.0, CH_2Cl_2)

Source of chirality: catalytic asymmetric hydroboration of styrene, followed by oxidation

Absolute configuration: *R*



E.e. = 90.5% [by converting into

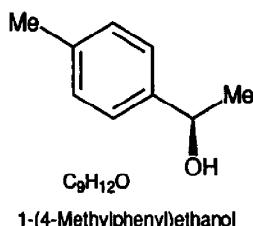


and HPLC with chiral stationary phase column, Sumipax OA-4100]

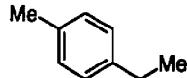
$[\alpha]_D^{21} +46.1$ (*c* 1.0, Et_2O)

Source of chirality: catalytic asymmetric hydroboration of 4-chlorostyrene, followed by oxidation

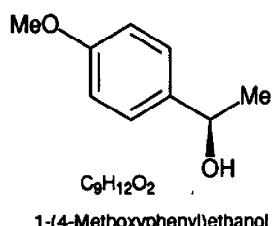
Absolute configuration: *R*



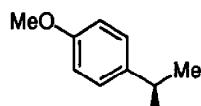
E.e. = 93.8% [by converting into

 $OCONHC_6H_3(NO_2)_2-3,5$

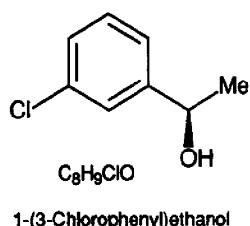
and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{25} +51.6 \quad (c\ 1.0, CHCl_3)$ Source of chirality: catalytic asymmetric hydroboration of 4-methylstyrene,
followed by oxidationAbsolute configuration: *R*

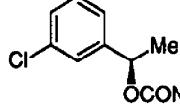
E.e. = 88.5% [by converting into

 $OCONHC_6H_3(NO_2)_2-3,5$

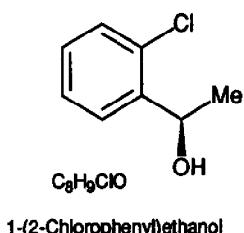
and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{20} +47.2 \quad (c\ 1.0, CHCl_3)$ Source of chirality: catalytic asymmetric hydroboration of 4-methoxystyrene,
followed by oxidationAbsolute configuration: *R*

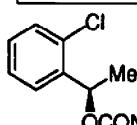
E.e. = 84.6% [by converting into

 $OCONHC_6H_3(NO_2)_2-3,5$

and HPLC with chiral stationary phase column, Sumipax OA-4100]

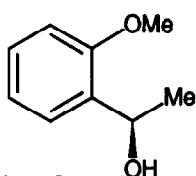
 $[\alpha]_D^{20} +36.7 \quad (c\ 1.0, CHCl_3)$ Source of chirality: catalytic asymmetric hydroboration of 3-chlorostyrene,
followed by oxidationAbsolute configuration: *R* (assigned by similarity in elution order in the HPLC analysis)

E.e. = 72.1% [by converting into

 $OCONHC_6H_3(NO_2)_2-3,5$

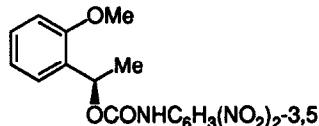
and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{20} +22.4 \quad (c\ 1.1, CHCl_3)$ Source of chirality: catalytic asymmetric hydroboration of 2-chlorostyrene,
followed by oxidationAbsolute configuration: *R* (assigned by similarity in elution order in the HPLC analysis)



1-(2-Methoxyphenyl)ethanol

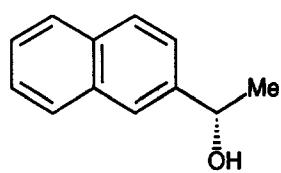
E.e. = 81.5% [by converting into



and HPLC with chiral stationary phase column, Sumipax OA-4100]

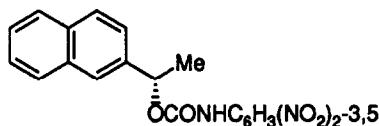
 $[\alpha]_D^{20} +48.9$ (c 1.1, toluene)

Source of chirality: catalytic asymmetric hydroboration of 2-methoxystyrene, followed by oxidation

Absolute configuration: *R*

1-(2-Naphthyl)ethanol

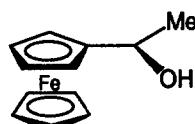
E.e. = 13.2% [by converting into



and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{20} -7.5$ (c 0.1, EtOH)

Source of chirality: catalytic asymmetric hydroboration of 2-vinylnaphthalene, followed by oxidation

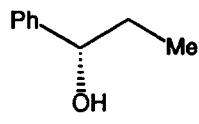
Absolute configuration: *S*

1-Ferrocenylethanol

E.e. = 58% [determined by comparison of optical rotation with that reported.]

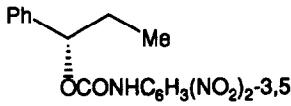
 $[\alpha]_D^{25} -17.0$ (c 1.1, benzene)

Source of chirality: catalytic asymmetric hydroboration of vinylferrocene, followed by oxidation

Absolute configuration: *R*

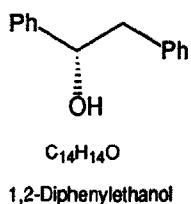
1-Phenyl-1-propanol

E.e. = 42.3% [by converting into

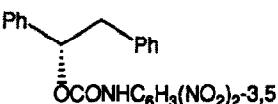


and HPLC with chiral stationary phase column, Sumipax OA-4100]

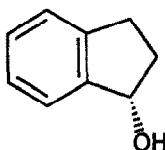
 $[\alpha]_D^{20} -20.6$ (c 1.0, CHCl₃)Source of chirality: catalytic asymmetric hydroboration of (*E*)-1-phenylpropene, followed by oxidationAbsolute configuration: *S*



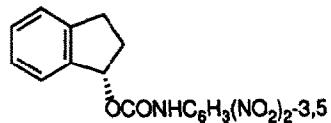
E.e. = 16.4% [by converting into



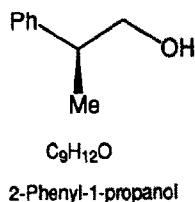
and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{23} +10.6$ ($c\ 1.0$, EtOH)Source of chirality: catalytic asymmetric hydroboration of (*E*)-stilbene,
followed by oxidationAbsolute configuration: *S*

E.e. = 13.1% [by converting into



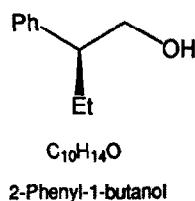
and HPLC with chiral stationary phase column, Sumipax OA-4100]

 $[\alpha]_D^{22.5} -3.81$ ($c\ 1.2$, $CHCl_3$)Source of chirality: catalytic asymmetric hydroboration of indene,
followed by oxidationAbsolute configuration: *S*

E.e. = 19.2% [by converting into



and HPLC with chiral stationary phase column, Sumipax OA-1100]

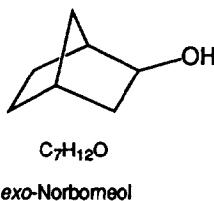
 $[\alpha]_D^{17} -4.0$ ($c\ 0.9$, benzene)Source of chirality: catalytic asymmetric hydroboration of 2-phenylpropene,
followed by oxidationAbsolute configuration: *S*

E.e. = 46.5% [by converting into



and HPLC with chiral stationary phase column, Sumipax OA-1100]

 $[\alpha]_D^{22} +8.0$ ($c\ 1.1$, EtOH)Source of chirality: catalytic asymmetric hydroboration of 2-phenyl-1-butene,
followed by oxidationAbsolute configuration: *S*



E.e. = 14.8% [by converting into

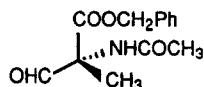


and HPLC with chiral stationary phase column, Sumipax OA-4100]

$[\alpha]_D^{25} -1.0$ ($c 1.1, CHCl_3$)

Source of chirality: catalytic asymmetric hydroboration of norbornene,
followed by oxidation

Absolute configuration: 1S, 2S, 4R



$C_{13}H_{15}NO_4$
Benzyl 2-formyl N-acetylalaninate

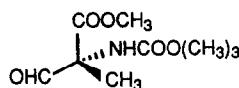
E.e. = 75% (by GLC with chiral capillary column)

$[\alpha]_D^{25} = +48 \pm 2$ ($c 2, \text{acetone}$) for the optically pure product

Source of chirality: asymmetric hydroformylation of benzyl
N-acetamidoacrylate

Absolute configuration : R

(assigned by correlation of configuration)



$C_{10}H_{17}NO_5$
Methyl 2-formyl-N-t-butyloxycarbonyl
alaninate

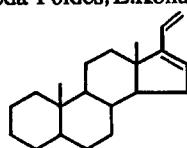
E.e. = 46% (by GLC with chiral capillary column)

$[\alpha]_D^{25} = +60 \pm 2$ ($c 2, \text{acetone}$) for the optically pure product

Source of chirality: asymmetric hydroformylation of methyl
N-t.butyloxycarbonylamidoacrylate

Absolute configuration : R

(assigned by correlation of configuration)

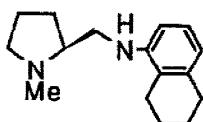


$C_{21}H_{32}$
16,20-Pregnadiene

Source of chirality: 17-iodo-androsta-16-ene

$[\alpha]_{546}^{20} +26.4$ ($c 0.53, \text{in } CHCl_3$)

mp 59°C



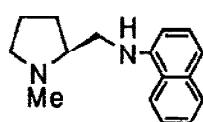
E. e. = Not determined
 $[\alpha]_D^{22} -27.1$ (c 0.88, EtOH)

Source of chirality: (S)-proline

Absolute configuration S

C₁₆H₂₄N₂

1-Methyl-2-[(N-1-(5,6,7,8-tetrahydronaphthyl)amino)methyl]pyrrolidine



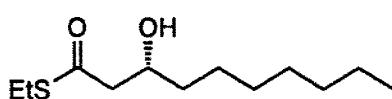
E. e. = Not determined
 $[\alpha]_D^{30} -35.6$ (c 1.05, EtOH)

Source of chirality: (S)-proline

Absolute configuration S

C₁₆H₂₀N₂

1-Methyl-2-[(N-1-naphthylamino)methyl]pyrrolidine



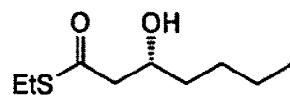
E. e. = 93% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{29} -30.0$ (c 2.99, benzene)

Source of chirality: asymm. synth. (aldol)

Absolute configuration R

C₁₂H₂₄O₂S

S-Ethyl 3-hydroxydecanethioate



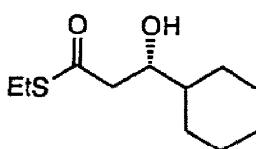
E. e. = 91% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{28} -37.9$ (c 4.03, benzene)

Source of chirality: asymm. synth. (aldol)

Absolute configuration R

C₉H₁₈O₂S

S-Ethyl 3-hydroxyheptanethioate

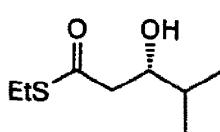
 $C_{11}H_{20}O_2S$

S-Ethyl 3-cyclohexyl-3-hydroxypropanethioate

E. e. = 92% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{32} -42.9$ (c 2.45, benzene)

Source of chirality: asymm. synth. (aldol)

Absolute configuration S

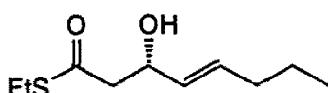
 $C_8H_{16}O_2S$

S-Ethyl 3-hydroxy-4-methylpentanethioate

E. e. = 90% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{28} -50.2$ (c 1.88, benzene)

Source of chirality: asymm. synth. (aldol)

Absolute configuration S

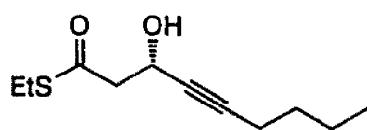
E. e. = 72% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{30} -43.2$ (c 1.87, benzene)

Source of chirality: asymm. synth. (aldol)

Absolute configuration S

 $C_{10}H_{18}O_2S$

S-Ethyl 3-hydroxy-trans-4-octenethioate

E. e. = 88% (by HPLC using Daicel Chiralcel OD)
 $[\alpha]_D^{27} -36.1$ (c 2.97, benzene)

Source of chirality: asymm. synth. (aldol)

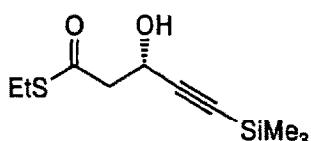
Absolute configuration S

 $C_{11}H_{18}O_2S$

S-Ethyl 3-hydroxy-4-nonynethioate

S. Kobayashi, M. Furuya, A. Ohtsubo and T. Mukaiyama

Tetrahedron: Asymmetry 1991, 2, 635



E. e. = 77% (by HPLC using Daicel Chiralcel AD)
[α]_D²⁹ -32.6 (c 1.80, benzene)

Source of chirality: asymm. synth. (aldol)

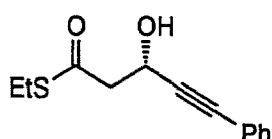
Absolute configuration S

C₁₀H₁₈O₂SSi

S-Ethyl 3-hydroxy-5-trimethylsilyl-4-pentynethioate

S. Kobayashi, M. Furuya, A. Ohtsubo and T. Mukaiyama

Tetrahedron: Asymmetry 1991, 2, 635



E. e. = 79% (by HPLC using Daicel Chiralcel AD)
[α]_D³⁰ -37.7 (c 5.70, benzene)

Source of chirality: asymm. synth. (aldol)

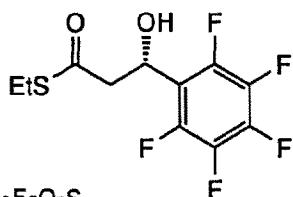
Absolute configuration S

C₁₃H₁₄O₂S

S-Ethyl 3-hydroxy-5-phenyl-4-pentynethioate

S. Kobayashi, M. Furuya, A. Ohtsubo and T. Mukaiyama

Tetrahedron: Asymmetry 1991, 2, 635



E. e. = 68% (by HPLC using Daicel Chiralcel OD)
[α]_D³⁰ +15.4 (c 1.15, benzene)

Source of chirality: asymm. synth. (aldol)

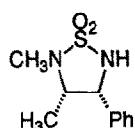
Absolute configuration S

C₁₁H₉F₅O₂S

S-Ethyl 3-hydroxy-3-pentafluorophenylpropanethioate

D. Sartor, J. Saffrich, G. Heimchen, C. J. Richards and
H. Lambert

Tetrahedron: Asymmetry 1991, 2, 639



E.e. = 100%

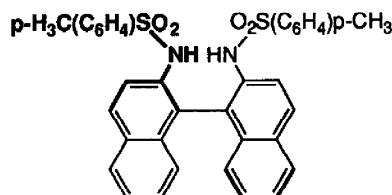
[α]_D²⁰ = + 18.6 (c = 3.5, CHCl₃)

Source of chirality: commercial (-)-ephedrine

C₁₀H₁₄N₂O₂S
(3R, 4S)-4,5-Dimethyl-1,1-dioxo-3-phenyl-1,2,5-thiadiazolidine

D. Sartor, J. Saffrich, G. Helmchen, C. J. Richards and H. Lambert

Tetrahedron: Asymmetry 1991, 2, 639



E.e. > 99%

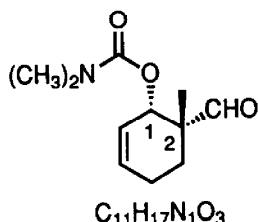
$[\alpha]_D^{20} = +70.5$ ($c = 2.2$, CHCl_3)

Source of chirality: commercial (+)-(R)-2,2'-diamino-1,1'-binaphthyl

$\text{C}_{34}\text{H}_{28}\text{N}_2\text{O}_4\text{S}_2$
(R)-N,N'-Ditosyl-2,2'-diamino-1,1'-binaphthyl

K. Mikami, M. Terada, Y. Motoyama and T. Nakai

Tetrahedron: Asymmetry 1991, 2, 643



E.e. = 85% (by LIS-NMR analysis)

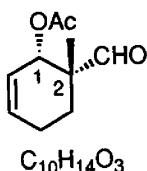
$[\alpha]_D^{25} = 165.3$ ($c 1.72$, CHCl_3) (99.4:0.6 *endo/exo*-mixture)

Source of chirality: Asymmetric Synthesis

Absolute configuration: 1S, 2R

K. Mikami, M. Terada, Y. Motoyama and T. Nakai

Tetrahedron: Asymmetry 1991, 2, 643



E.e. = 78% (by LIS-NMR analysis)

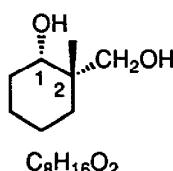
$[\alpha]_D^{25} = 168.4$ ($c 0.98$, CHCl_3) (97.3 *endo/exo*-mixture)

Source of chirality: Asymmetric Synthesis

Absolute configuration: 1S, 2R

K. Mikami, M. Terada, Y. Motoyama and T. Nakai

Tetrahedron: Asymmetry 1991, 2, 643



E.e. = 85% (by NMR analysis after conversion to MTPA ester)

$[\alpha]_D^{23} = 14.8$ ($c 2.95$, CHCl_3)

Source of chirality: Asymmetric Synthesis

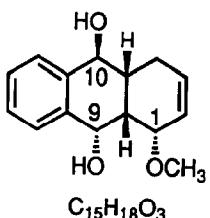
Absolute configuration: 1S, 2S



E.e. = 85% (Not isolable. The % ee was determined after reduction)

Source of chirality: Asymmetric Synthesis

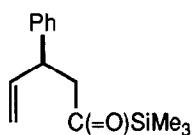
Absolute configuration: 1(S), 4a(R), 9a(R) (supposed on the basis of analogy with general method)



E.e. = 85% (by NMR analysis after conversion to MTPA ester)

Source of chirality: Asymmetric Synthesis

Absolute configuration: 1(S), 4a(R), 9(S), 9a(S), 10(S) (supposed on the basis of analogy with general method)



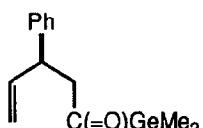
E.e. = 88%

$[\alpha]_D^{24} -38.5$ (*c* 0.5, CHCl₃)

Source of chirality: asymmetric Claisen rearrangement

Absolute configuration: 3*S*

$C_{14}H_{20}OSi$
(*S*)-3-phenyl-1-(trimethylsilyl)-4-penten-1-one



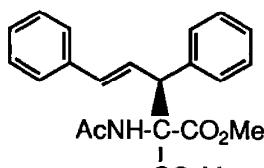
E.e. = 91%

$[\alpha]_D^{24} -23.7$ (*c* 0.5, CHCl₃)

Source of chirality: asymmetric Claisen rearrangement

Absolute configuration: 3*S*

$C_{14}H_{20}OGe$
(*S*)-3-phenyl-1-(trimethylgermyl)-4-penten-1-one



Dimethyl 2-acetamido-2-((E)-1,3-diphenyl-2-propenyl)malonate

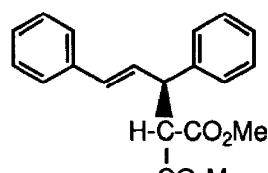
E.e. = 94% (by ^1H NMR)

$[\alpha]_D^{20} = -51$ ($c = 0.6$, EtOH)

Circular dichroism : $\lambda_{\text{EtOH}}^{\text{max}} (\Delta\epsilon) = 288 \text{ nm} (-0.18)$

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : S (assigned by chemical transformation from dimethyl 2-((E)-1,3-diphenyl-2-propenyl)malonate)



Dimethyl 2-((E)-1,3-diphenyl-2-propenyl)malonate

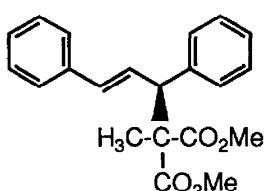
E.e. = 90% (by ^1H NMR)

$[\alpha]_D^{20} = +18$ ($c = 0.6$, EtOH)

Circular dichroism : $\lambda_{\text{EtOH}}^{\text{max}} (\Delta\epsilon) = 292 \text{ nm} (-0.10)$

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : R (ref. Hayashi, T.; Yamamoto, A.; Hagiwara, T.; Ito, Y. *Tetrahedron Lett.* 1986, 27, 191.)



Dimethyl 2-methyl-2-((E)-1,3-diphenyl-2-propenyl)malonate

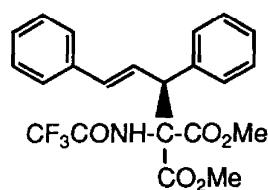
E.e. = 80% (by ^1H NMR)

$[\alpha]_D^{20} = -37$ ($c = 0.6$, EtOH)

Circular dichroism : $\lambda_{\text{EtOH}}^{\text{max}} (\Delta\epsilon) = 295 \text{ nm} (-0.15)$

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : R



Dimethyl 2-trifluoroacetamido-2-((E)-1,3-diphenyl-2-propenyl)malonate

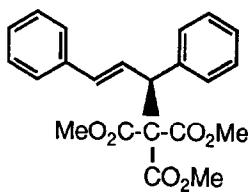
E.e. = 78% (by ^1H NMR)

$[\alpha]_D^{20} = -41$ ($c = 0.6$, EtOH)

Circular dichroism : $\lambda_{\text{EtOH}}^{\text{max}} (\Delta\epsilon) = 286 \text{ nm} (-0.43)$

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : S



Dimethyl 2-methoxycarbonyl-2-((E)-1,3-diphenyl-2-propenyl)malonate

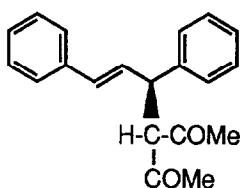
E.e. = 81% (by 1H NMR)

$[\alpha]_D^{20} = -23$ ($c = 0.6$, EtOH)

Circular dichroism : $\lambda_{\text{max}}^{\text{EtOH}} = 290 \text{ nm}$ ($\Delta\varepsilon = -0.36$)

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : S



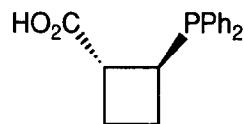
3-((E)-1,3-diphenyl-2-propenyl)-2,4-pentanedione

E.e. = 90% (by 1H NMR)

$[\alpha]_D^{20} = -10$ ($c = 0.6$, EtOH)

Source of chirality : Asymm. synth. with palladium complex

Absolute configuration : R (ref. Hayashi, T.; Yamamoto, A.; Hagihara, T.; Ito, Y. *Tetrahedron Lett.* 1986, 27, 191.)



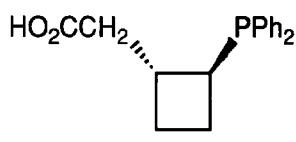
trans-(2-Diphenylphosphino)cyclobutanecarboxylic acid

(-) -Form: optical purity = 96 %ee

$[\alpha]_D = -90.1$ ($c 6.3$, CH_2Cl_2)

(+) -Form: optical purity = 92 %ee

$[\alpha]_D = 86.5$ ($c 2.2$, CH_2Cl_2)



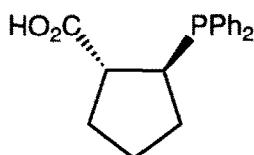
trans-[2-(Diphenylphosphino)cyclobutyl]acetic acid

(+) -Form: optical purity => 99 %ee

$[\alpha]_D = 24.1$ ($c 1.6$, CH_2Cl_2)

Y. Okada, T. Minami, Y. Umezu, S. Nishikawa, R. Mori, and
Y. Nakayama

Tetrahedron: Asymmetry 1991, 2, 667



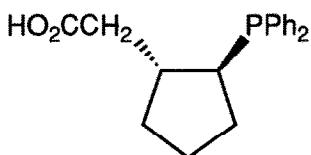
(-) -Form: optical purity = 83 %ee
[α]_D = -29.6 (c 1.8, CH₂Cl₂)

(+) -Form: optical purity = 95 %ee
[α]_D = 31.4 (c 1.0, CH₂Cl₂)

trans-(2-Diphenylphosphino)cyclopentanecarboxylic acid

Y. Okada, T. Minami, Y. Umezu, S. Nishikawa, R. Mori, and
Y. Nakayama

Tetrahedron: Asymmetry 1991, 2, 667

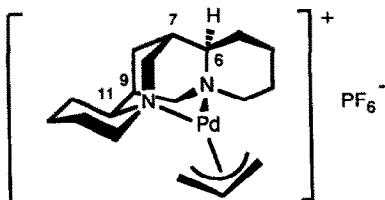


(-) -Form: optical purity = 94 %ee
[α]_D = -2.69 (c 1.0, CH₂Cl₂)
(+) -Form: optical purity = 97 %ee
[α]_D = 6.79 (c 1.0, CH₂Cl₂)

trans-[2-(Diphenylphosphino)cyclopentyl]acetic acid

Antonio Togni

Tetrahedron: Asymmetry 1991, 2, 683

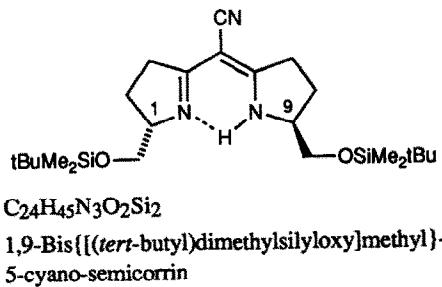


(η³-Allyl)(Sparteine)palladium(II) Hexafluorophosphate
C₁₈H₃₁F₆N₂PPd

E.e. = 100 %
[α]_D²² = -67 (c = 1.035, CH₂Cl₂)
Source of chirality : natural
Absolute configuration (of natural (-)-sparteine):
6R, 7S, 9S, 11S

P. von Matt and A. Pfaltz

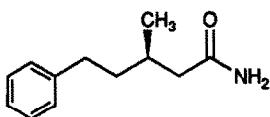
Tetrahedron: Asymmetry 1991, 2, 691



[α]_D = -64.7 (c 1.0, CHCl₃, 23 °C)

Source of chirality: synthesis from L-pyroglutamic acid

Absolute configuration 1S, 9S
(based on configuration of pyroglutamic acid)

 $C_{12}H_{17}NO$

3-Methyl-5-phenylpentanamide

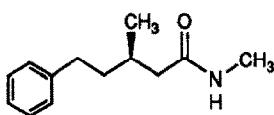
E.e. = 95.1 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide)

$[\alpha]_D = +17.6$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *R*

(assigned by optical rotation according to the literature)

 $C_{13}H_{19}NO$

N,3-Dimethyl-5-phenylpentanamide

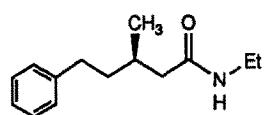
E.e. = 98.7 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide)

$[\alpha]_D = +18.4$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *R*

(assigned by optical rotation according to the literature)

 $C_{14}H_{21}NO$

N-Ethyl-3-methyl-5-phenylpentanamide

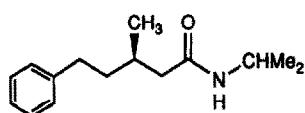
E.e. = 94.8 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide)

$[\alpha]_D = +14.7$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *R*

(assigned by optical rotation according to the literature)

 $C_{15}H_{23}NO$

N-Isopropyl-3-methyl-5-phenylpentanamide

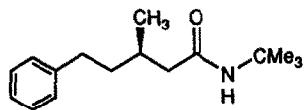
E.e. = 93.5 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide)

$[\alpha]_D = +9.6$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *R*

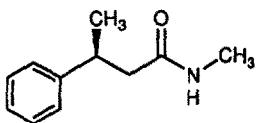
(assigned by optical rotation according to the literature)

 $C_{16}H_{25}NO$ N-*tert*-Butyl-3-methyl-5-phenylpentanamideE.e. = 92.0 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide) $[\alpha]_D = +8.9$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *R*

(assigned by optical rotation according to the literature)

 $C_{11}H_{15}NO$

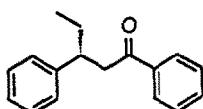
N-Methyl-3-phenylbutanamide

E.e. = 92.3 % (by HPLC analysis of the corresponding (*R*)-1-naphthyl)ethylamide) $[\alpha]_D = +38.2$ (c 1.0, CHCl₃, 23 °C)

Source of chirality: asymmetric synthesis

Absolute configuration *S*

(assigned by optical rotation according to the literature)

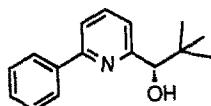
 $C_{17}H_{18}O$

1,3-Diphenylpentanone

Ee = 86% (by HPLC analysis)

Absolute configuration: R

Source of chirality: asymmetric synthesis



Ee = 98% (by HPLC analysis)

Absolute configuration: S

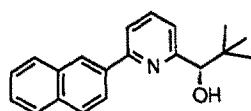
Source of chirality: asymmetric synthesis

 $C_{16}H_{19}NO$

2,2-Dimethyl-1-(6-phenyl-pyridin-2-yl)propanol

C. Bolm

Tetrahedron: Asymmetry 1991, 2, 701



Ee = 98% (by HPLC analysis)

Absolute configuration: S

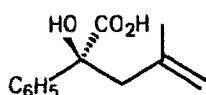
Source of chirality: asymmetric synthesis

C₂₀H₂₁NO

2,2-Dimethyl-1-[6-(naphth-2-yl)-pyridin-2-yl]propanol

H. Moorlag and R.M. Kellogg

Tetrahedron: Asymmetry 1991, 2, 705



C₁₂H₁₄O₃

2-Hydroxy-2-phenyl-4-methyl-4-pentenoic acid

ee ≥ 98% (by ¹H NMR with (S)-2-chloropropanoyl chloride)

[α]_D²⁰ = -27.0 (c 1, CHCl₃)

Source of chirality: enzymatic resolution

Absolute configuration: S

(assigned by chemical correlation)

H.U. Blaser*, S.K. Boyer and U. Pittelkow

Tetrahedron: Asymmetry 1991, 2, 721

e.e. = 50% (by HPLC on tribenzoylcellulose)

[α]_D²⁰ = 370 ± 20 (c=0.5, MeOH)

Source of chirality: cinchonine in the catalytic system

Pd/BaSO₄ - cinchonine

Absolute configuration: R

3(R)-3-chloro-2,3,4,5-tetrahydro-1H-[1]benzazepin-2-one